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(1RS,2SR,6SR,8SR)-2,6-Dimethyl-8-(phenyl-sulfonyl)bicyclo[4.2.0]octan-1-ol

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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(\text{C-C}) = 0.004 \text{ Å}$ R factor = 0.047 wR factor = 0.131Data-to-parameter ratio = 19.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound, $C_{16}H_{22}O_3S$, exhibits a conformational arrangement that permits intramolecular $O-H\cdots O-S$ hydrogen bonds. Intermolecular bifurcated hydrogen bonding between the hydroxyl H and the sulfone O atoms, observed in the unmethylated derivative, is disrupted by the presence of the methyl group at the 6-position.

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Comment

We have recently shown that a novel cyclization reaction between the lithium enolates of simple unfunctionalized ketones and phenyl vinyl sulfoxide provides a simple and convenient route to the preparation of fused carbocyclic ring systems bearing a bridgehead hydroxyl group (Loughlin & McCleary, 2003; Loughlin, Rowen & Healy, 2002). In the current study, reaction of the lithium enolate of 2,6-dimethylcyclohexan-1-one (obtained from lithium diisopropylamide, LDA) with phenyl vinyl sulfoxide and subsequent oxidation with *m*-chloroperoxybenzoic acid (*m*-CPBA) generated a novel dimethylbicyclo[4.2.0]octan-1-ol, (I).

A bicyclo[4.2.0]octan-1-ol with a methyl bridgehead substituent can be perceived as a key structural component of, for example, natural products such as melleolide K (Momose et al., 2000) and cyclobutatusin (Zelnik et al., 1977; Wang et al., 1974) and as a synthetic intermediate in natural product synthesis (Morisaki et al., 1985, Ayer et al., 1981). Under the present unoptimized reaction conditions, (I) was formed as the major bicyclo [4.2.0] octan-1-ol isomer in a 22:78 ratio of (I):(II) (Loughlin, McCleary & Healy, 2002) and (III) from 2,6-dimethylcyclohexan-1-one and phenyl vinyl sulfoxide with less than 5 percent of other products observed (see Scheme). Here we report the synthesis, isolation and solid-state structural characterization of the novel bicyclo[4.2.0]octan-1-ol (I), Fig. 1. The S-O2 and S-O3 bonds are directed away from the bicyclo[4.2.0]alkan-1-ol ring, whereas the phenyl group is oriented towards this ring. Intramolecular hydrogen-bonding

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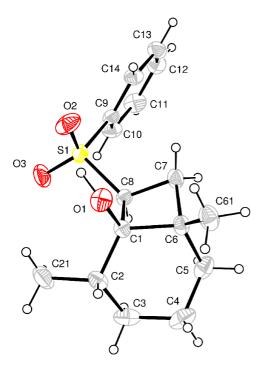


Figure 1 *ORTEP*-3 (Farrugia, 1997) plot, showing the atomic numbering scheme for the molecule of (I). Displacement ellipsoids for non-H atoms are drawn at the 30% probability level.

interactions are observed between the hydroxyl H atom and the sulfone O atom O2, with an O1···O2 distance of 2.765 (3) Å and an estimated O—H···O angle of 153°. The six-membered ring displays a pseudo-chair conformation. In the previously reported 8-(phenylsulfonyl)bicyclo[4.2.0]octan-1-ol, which lacks the 2,6-dimethyl groups of compound (I), the orientation of the six-membered ring allows intermolecular bifurcated hydrogen bonding between the hydroxyl H atom and the sulfone O atoms (Healy *et al.*, 2002). However, in compound (I) the presence of the methyl group in the C6 position now disrupts this and intermolecular hydrogen bonding is not observed in the solid-state structure of (I).

Experimental

2,6-Dimethylcyclohexanone (0.55 ml, 3.96 mmole), lithium diisopropylamide (1.55 M, 2.5 6 ml, 3.96 mmole) in THF, and phenyl vinyl sulfoxide (0.5 5 ml, 3.96 mmol) were reacted at 263 K with a 5 min reaction time, worked up and subsequently oxidized with m-CPBA (1 equiv) in chloroform (50 ml) as described elsewhere (Loughlin, Rowen & Healy, 2002). After silica chromatography (hexane:ethyl acetate, 80:20), compound (I) was obtained (197 mg, 17%) in conjunction with monoalkylated 2,6-dimethylcyclohexanone, compounds (II) and (III) (577 mg, 53%) and other minor products (3%). An analytically pure sample of compound (I) was obtained by semi-preparative HPLC (hexane–ethyl acetate, 90:10, retention time 17.4 min, 3 ml min $^{-1}$). Colourless crystals of compound (I) (m.p. 381.5–382.1 K) were isolated by slow evaporation of a hexane:ethyl acetate (90:10) solution of (I). Analysis found: C 65.34, H 7.64, O 16.36%; calculated for $C_{16}H_{22}SO_3$: C 65.27, H 7.53, O 16.30%.

Crystal data

$D_x = 1.274 \text{ Mg m}^{-3}$
Mo K α radiation
Cell parameters from 25
reflections
$\theta = 12.6 – 17.4^{\circ}$
$\mu = 0.22 \text{ mm}^{-1}$
T = 295 K
Prism, colourless
$0.40\times0.35\times0.20~\text{mm}$

Data collection

Rigaku AFC-7R diffractometer	$\theta_{\rm max} = 27.5^{\circ}$
ω –2 θ scans	$h = -6 \rightarrow 14$
Absorption correction: none	$k = -23 \rightarrow 10$
4336 measured reflections	$l = -10 \rightarrow 9$
3518 independent reflections	3 standard reflections
1904 reflections with $I > 2\sigma(I)$	every 150 reflections
$R_{\rm int} = 0.053$	intensity decay: 0.8%

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0382P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.047$	+ 0.7618P]
$wR(F^2) = 0.131$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{\rm max} = 0.001$
3518 reflections	$\Delta \rho_{\text{max}} = 0.23 \text{ e Å}^{-3}$
182 parameters	$\Delta \rho_{\min} = -0.27 \text{ e Å}^{-3}$
H-atom parameters constrained	

Table 1 Selected geometric parameters (Å, °).

S1-O2	1.445 (2)	S1-C9	1.768 (3)
S1-O3	1.432 (3)	O1-C1	1.425 (3)
S1-C8	1.770(2)		
O2-S1-O3	118.11 (14)	O1-C1-C8	112.01 (18)
O2 - S1 - C8	108.50 (12)	O1 - C1 - C2	108.1(2)
O2 - S1 - C9	107.31 (12)	S1-C8-C1	121.69 (16)
O3-S1-C8	110.16 (13)	S1-C8-C7	117.36 (18)
O3-S1-C9	108.45 (13)	S1-C9-C10	119.7 (2)
C8-S1-C9	103.25 (11)	S1-C9-C14	118.67 (19)
O1-C1-C6	109.7 (2)		

H atoms were positioned geometrically (C-H = 0.95, O-H = 0.96 Å) and refined as riding, with $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm parent\ atom})$.

Data collection: MSC/AFC-7 Diffractometer Control Software for Windows (Molecular Structure Corporation, 1999); cell refinement: MSC/AFC-7 Diffractometer Control Software for Windows; data reduction: TEXSAN for Windows (Molecular Structure Corporation, 1997–2001); program(s) used to solve structure: TEXSAN for Windows; program(s) used to refine structure: TEXSAN for Windows and SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2001) and ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: TEXSAN for Windows and PLATON.

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